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USASRDL Technical Report 2196

PROPERTIES OF HOT-PRESSED BARIUM TITANATE

by

Arthur Brown

and

Robert Fischer



April 1961



5. U.S. ARMY SIGNAL RESEARCH AND DEVELOPMENT LABORATORY FORT MONMOUTH, N. J.

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DA TASK NR. 3A99-15-001-01

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PROPERTIES OF HOT-PRESSED BARIUM TITANATE

INTRODUCTION

The objective of the program described was to develop a technique for producing a r dense, hot-pressed barium titanate sample having an average grain size of less than one and to compare its electrical properties with those of conventionally-fired hari in titana

Barium titanate had been hot-pressed by others using graphite dies. Hot-pressing ir results in reduced barium titanate ceramic which, upon subsequent oxidation, was subjegrain growth. Hot-pressing of cadmium niobate in ceramic dies had previously been den at USASRDL by DeBretteville, et al, 1 and barium titanate had been hot-pressed and son properties reported by Tennery. 2

Further development of the hot-pressing technique has enabled the authors to production reduced samples of extremely fine grain structure whose electrical properties are of

METHOD

The hot-pressing apparatus consists of an induction furnace used in conjunction wit and die assembly. An over-all view of the assembled apparatus is shown in Figure 1.

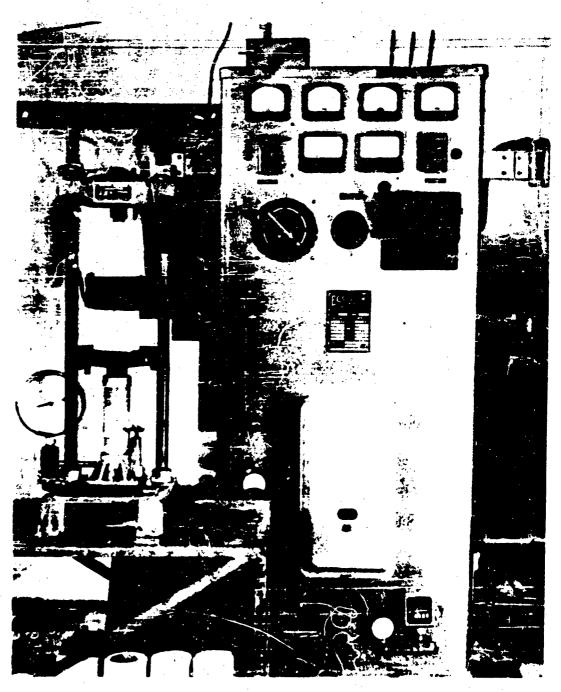
The induction power supply is a 20-kw, 10-kc motor generator, which allows rapid he with good control. The furnace, which was assembled on the platen of the press, was a insulating fire brick cut to fit inside the induction coil and around the die assembly. A tional, manually-operated hydraulic laboratory press was used. The furnace and die assare shown in Figure 2

The ceramic dies which have proven most effective for our work with barium titanate 100%, stabilized zirconia made slightly porous to improve thermal shock resistance. Ot materials such as alumina and alumina porcelains can also be used with some degree of A stainless steel bushing was shrunk onto the ceramic sleeve to serve as the susceptor add mechanical support.

The method used to prevent adherence of the material being pressed to the ceramic content the sample in a relatively coarse zirconia powder (TAM No. 45006). This method sticking to the die and allowed the sample to be ejected easily while hot. The amount c that diffused into the sample during firing has not been determined, but the layer of zirce that adhered to the surface of the sample was easily removed by grinding.

PROCEDURE

A sample disc of harium titanate 1./5 inch smaller in diameter than the zirconia die v preformed at 10,000 psi with no binder. This preformed disc was loaded into the hot-pre in such a manner as to he completely encapsulated in zirconia placing sand. The loader was aligned in the induction coil, and a pressure of 5000 psi was applied and maintainer out the heating cycle. The rate of heating was controlled in such a manner that's tempe 2000 F was reached in ten minutes. The sample was held of this temperature and press twenty minutes and then quickly ejected from the die and huried in vermiculity to cool, cooling, the sample was removed from its zirconia casing, ground to the desired shape, a polished.



M-58-315

FIGI HOT-PRESSING APPARATUS

HYDRAULIC PRESS HEAD-

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HYDRAULIC PRESS HEAD

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Other combinations of pressure, time, and temperature of firing were investigated; however, to such their 50% psi lead to rapid deterioration of the dies, and a significantly lower temperature or shorter time does not yield a fully vitrified body.

CHARACTERISTICS OF HOT-PRESSED Ballio,

Density and Microstructure

The polished samples had a yellowish color that is characteristic of nonreduced barium titanale. Fractured surfaces were glassy, an indication of the high density and fine grain size. Density measurements, made by displacement, averaged 5.9 gm/cc, some samples measuring as high as 6.0 gm/cc. When eliamined under 500 X magnification, polished samples exhibited necessity and structure before or after etching.

After careful polishing, samples were etched in a solution of 0.5% HF and 1.0% HNO3, and plantium shadowed carbon backed replicas were prepared and examined with the electron ruleroscope at magnifications up to 20,000 N. It was four dithat due to the small grain size a solistic torpoolishing and etching technique could not be developed.

Fractured surfaces were realizated to all electric macrographs are of unetched fractured surfaces. The grain size in the hotermissed barium titanate was found to be less than 1 micron thigure 3... Figure 4 illustrates the large grain size observed in a fractogram of kiln-fired barium titanate.

The fine microstructure of served in the hot-pressed samples is attributed to the fine particle size of the starting material, and to the predominance of plastic flow as the nechalism of sintering or der these conditions. The average particle size of the barrum titanate used (TAM FT grade) was 0.5 micron. These particles were found to be applomerated in such a manner that they indicated a particle size distribution of 1 to 3 microns by sedimentation methods. Electron micrographs showed the actual particles to be 1 micron and less (see Figure 5), essentially the same size as those appearing in the hot-pressed samples. This indicates that there was no significant grain growth during densification and that hot-pressing offers a method to be used to control grain size.

Electrical Characteristics

The electrical properties of hot-pressed partum fittinate samples prepared as above over quite striking when compared to those of componitionally prepared samples. The dielectric strength has been found to be 5-0 volts per rule as compared with values of 100 to 200 volts per rule usually observed in conventionally processed nature titations samples 1.110 inch three.

Figure 6 illustrates the differences observed in measurements of the dielectric constant and accordance factor with temperature. The reduction in the peak of the dielectric constant in the vicinity of the Cane temperature is attributed to the fine stant size. At from temperature the dielectric constant is allowed as a unit is independent of the entendency was found from the open to the dielectric constant is allowed as a prediction. No dependence of K on trequency was found from the open to the dielectric formation of the samples were descred by hearing to dielectric conventional and octobressed BaTitle. The samples were descred by hearing to dielectric formation as in the first temperature of temperature the dielectric constant has their recorded to be dielectric formation of the temperature of the dielectric constant has their recorded to be dielectric formation of the temperature as a constant, the results of this means formation of apply that the localisms in the hospitassed material where result very statue, if The explored stability has not been as they expressed in converting at K on a potter.

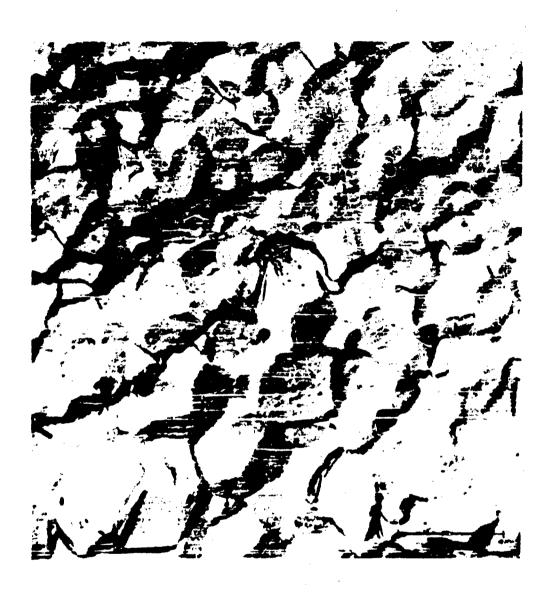


FIG 3 FRACTURED SURFACE OF HOT-PRESSED BATIO:

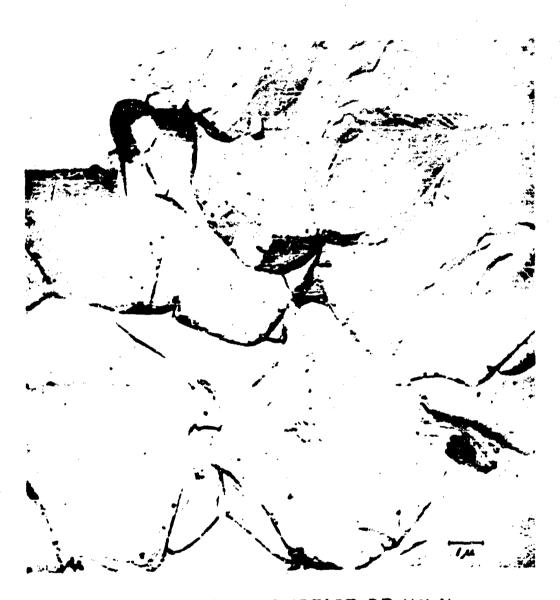


FIG 4 FRACTURD SURFACE OF KILN-FIRED BATIO:

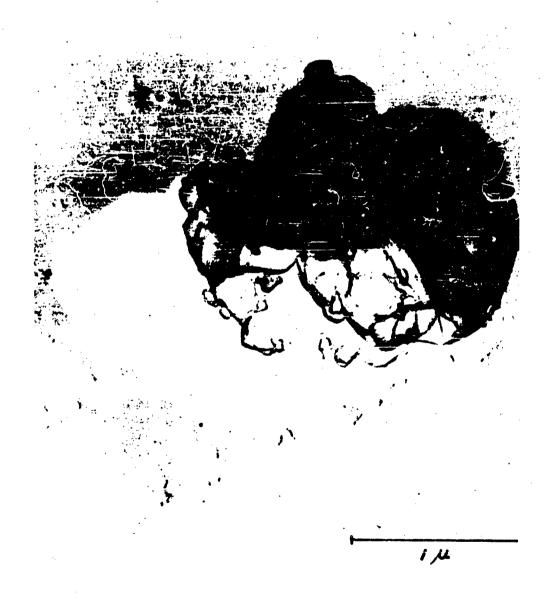


FIG 5 PARTICLE REPLICA OF BATIOS POWDER

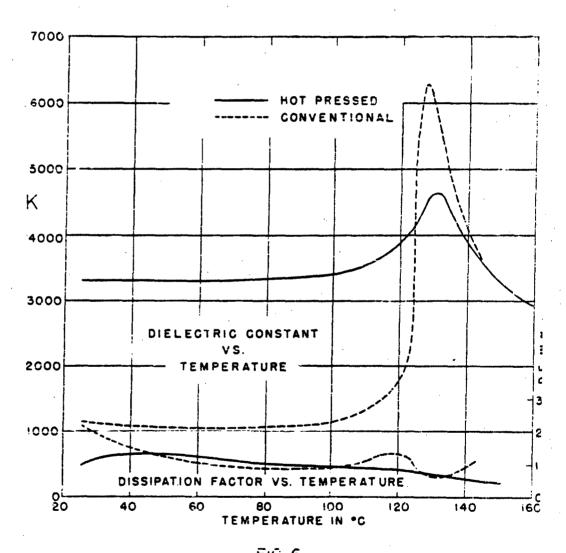
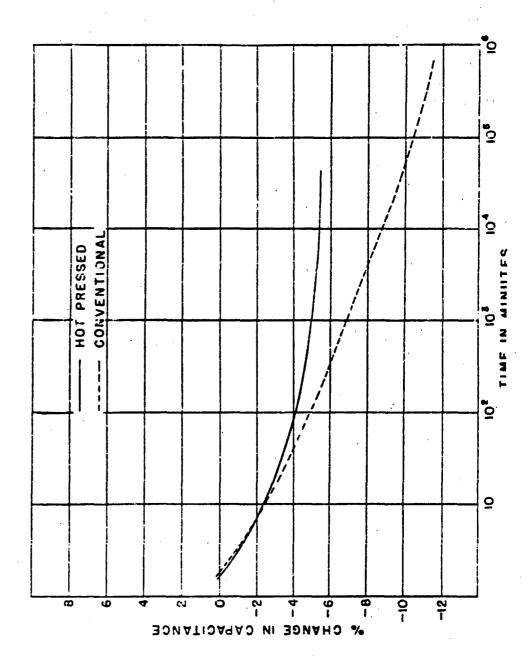


FIG 6

DIELECTRIC PROPERTIES OF TWO TYPES OF BARIUM TITANATE



An arrent was made to pole hor-pressed samples of BaTiO₃ by applying a high DC field at from temperature or while cooling from 120°C, but little or no piezoelectric response was observed.

CONCLUSIONS:

The hor-pressing technique described in this report produced a dense, nonreduced barium transfer sample with no apparent grain growth. The limiting factor in producing a fine-grained BaTiO, body was in the particle size of the raw materials. A project has been initiated to produce BaTiO, powder with an average particle size of less than 0.1 micron.

the dielectric properties of these materials were greatly enhanced by small crystallite size, but the piezoelectric response was almost completely eliminated. These results show the significance of the effect of density and grain size on the ferroelectric properties of barium titanate and indicate that an extension of this study over a wider range of particle sizes, particularly below 0.1 micron, might be expected to produce new or improved electrical properties in ferroelectric materials.

ACKNOWLEDGMENTS

The authors are grateful to John Charlton and Anne Dunlap of the Inorganic Dielectries Team for compiling all the electrical measurements; to Sam DiVita and Ronald Brandmayr of the Inorganic Dielectrics Team and Nelson Terbure, former Chief of the Inorganic Dielectrics Team, for their guidance and assistance in preparing this report; and to Charles Cook, Jr., of the Exploratory Research Division "E" for the instructions in electron microscopy.

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